#### Science at the heart of medicine

### **Hazard Communication Training**

#### Albert Einstein College of Medicine Department of Environmental Health & Safety http://www.einsteinmed.edu/ehs

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# **Training Outline**

- Federal Laboratory Standards
- Globally Harmonized System of Classification and Labeling of Chemicals (GHS)
- Potential Hazards at Einstein
- Safety Data Sheets (SDS)
- Chemical Toxicology
- Personal Protective Equipment (PPE)
- Chemical Storage/Disposal
- Emergency Procedures

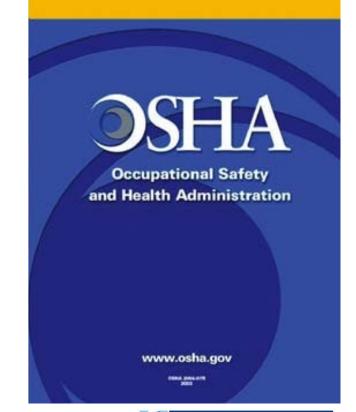


### What is Hazard Communication?

Defined under two Occupational Safety and Health Administration (OSHA) regulations:

Hazard Communication Standard (HCS) [29 CFR1910.1200]

Occupational Exposure to Hazardous Chemicals in Laboratories (29 CFR 1910.1450)





# Hazard Communication Goal



"Protect people from injuries and illnesses associated with using hazardous chemicals in the workplace".



# **Hazard Communication Standard**

#### Purpose:

- To ensure that the hazards of all chemicals produced or imported are classified and that the information is transmitted to employees.
- To provide employees with the necessary tools to protect themselves and their coworkers.
- HCS is aligned with the United Nations Globally Harmonized System of Classification and Labeling of Chemicals (GHS).





# **Hazard Communication Standard**

Accomplished via:

- >Written Hazard Communication Plan
- Chemical inventory
- Employee training
- Safety Data Sheets (SDS)
- Personal Protective Equipment (PPE)
- Engineering controls
- Container labeling & warnings



# What is the "Globally Harmonized System of Classification of Labeling of Chemicals" (GHS)?

 An international system for standardizing and harmonizing the classification of health, physical, and environmental hazards and labeling of chemicals including mixtures.

Globally Harmonized System of Classification and Labelling of Chemicals (GHS)

Ninth revised edition



#### What is the "GHS"? (Continued)

 Information about identities and hazards of the chemicals must be available and understandable to workers.





#### Globally Harmonized System of Classification of Labeling of Chemicals (GHS)

- Implemented to harmonize and improve safety information and to facilitate international trade through compatible classification methods, SDS and labels.
- Accomplished via:
  - Defining health, physical and environmental hazards of chemicals
  - Creating classification processes that use available data on chemicals for comparison with the defined hazard criteria
  - Communicating hazards and protective measures on labels and SDS.



# **HCS and Adoption of GHS**

#### Major changes to the HCS are:

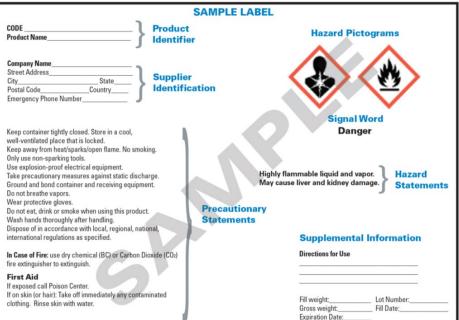
- Hazard Classification: Provides specific criteria for classification of health and physical hazards, as well as classification of mixtures.
- Labels: Chemical manufacturers and importers are required to provide a label that must include a harmonized signal word, pictogram, and hazard statement for each hazard class and category, and a Precautionary statement.
- > Safety Data Sheets: Now have a 16-section format.
- Information and Training: Employers are required to train workers on the new label elements and SDS to facilitate recognition and understanding.



#### HCS Adoption of GHS – Labels & Warnings

#### Labels and Warnings

- Each container of hazardous chemicals leaving the chemical manufacturer must be labeled, tagged or marked.
- Workplace labeling: Check secondary container labels for consistency.
- Label identities should link to the SDS & chemical inventories.





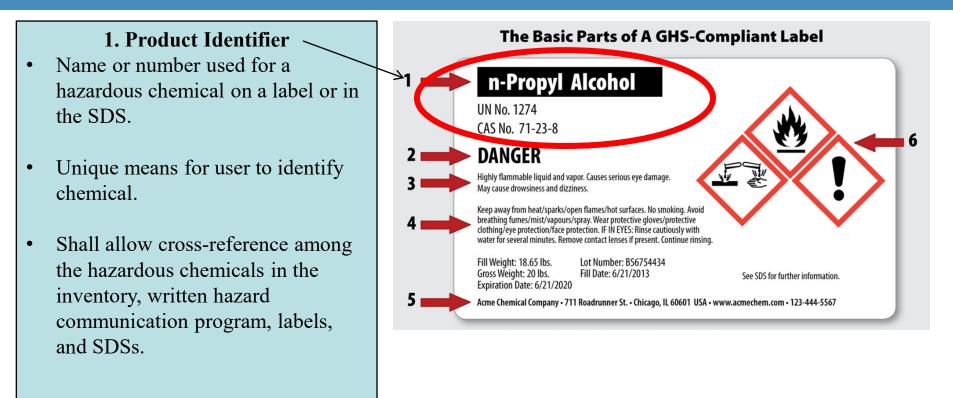
# **GHS** – Labels



Provide employees with the specific information regarding the physical and health hazards of the hazardous chemicals.



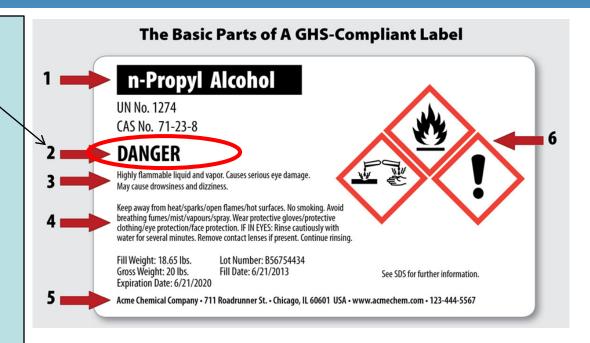






2. Signal Word
Used to indicate the relative level of severity of hazard and alert the reader to a potential hazard on the label.

- Signal words used are
   "Danger" and "Warning."
  - ➢ Danger − More severe
  - ➢ Warning − Less severe





#### **3.** Hazard Statements

- Phrase assigned to each product
- Means the nature of the physical or health hazards of the product, including where appropriate, the degree of hazard

#### ➢ Physical

- Example: H200 unstable explosive Highly Flammable, Flammable solid
- ≻Health
  - Example: H300 fatal if swallowed Causes serious eye damage, carcinogen, oral acute toxicity, etc.
- ➢ Environmental
  - Example: H400 Very toxic to aquatic life





#### 4. Precautionary Statements

Phase describing the recommended measures to minimize or prevent adverse effects resulting from exposure to a hazardous chemical, or improper storage or handling.

Precautionary Statements:

- 1. Prevention Ex. P235 "Keep cool"
- Response in case of accidental spills or exposure- Ex. P380 Evacuate Area
- Storage Ex. P403 Store in a well-ventilated space
- Disposal dispose in approved waste disposal facility; contents, container
- 5. General -Ex. P102 Keep out of reach of children







Provides name, address, and telephone number of the manufacturer or supplier

(Some may have the company website)





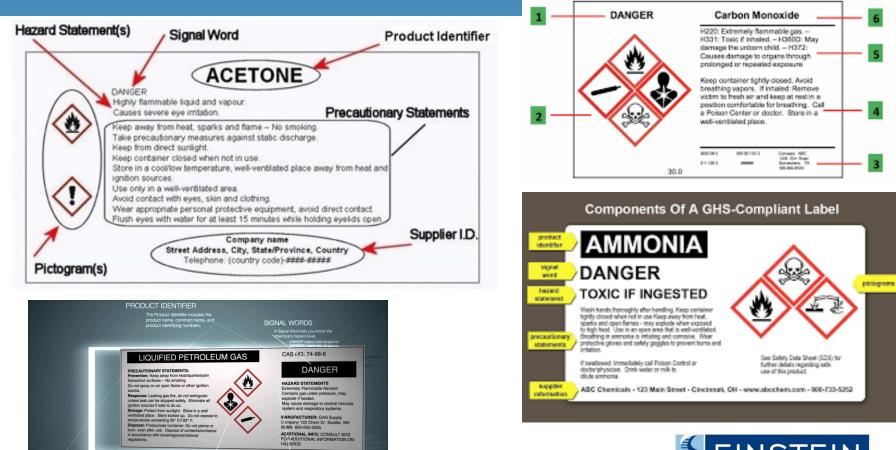
### GHS – Labels & Warnings (Examples)

ARD STATEMENTS

HAZARD PICTOGRAMS

PRECAUTIONARY STATEMENTS

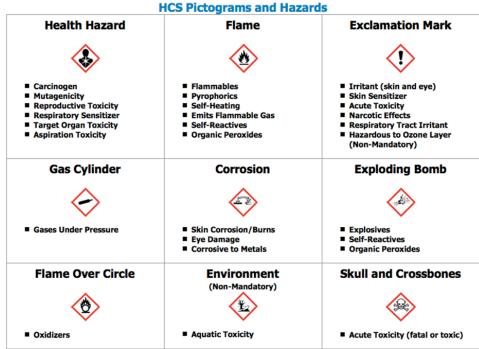
Hecommended measures that must be taken to minimize or prevent adver-





#### Pictogram

- A symbol plus other graphic elements, such as a border, background pattern, or color that is intended to convey specific information about the hazards of a chemical.
- Eight are mandatory and designated for application to hazard category.

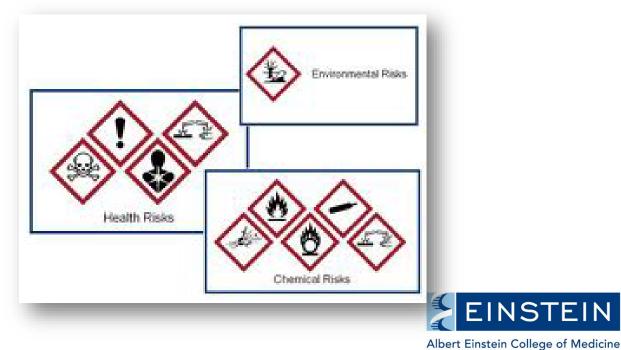




# HCS Adoption of GHS – Pictograms

- Health Risk
  - Severe Toxics
  - > Acute Toxics
  - Health Dangers
  - > Corrosives
- Chemical Risk
  - Explosives
  - ➢ Flammables
  - > Oxidizers
  - Gases under pressure
  - Corrosives

- Environmental Hazard Class
  - OSHA does not regulate the Environmental Hazard Class; however, the EPA is expected to incorporate this element of GHS into their standards.



# HCS Adoption of GHS – Workplace Labeling

- Secondary Container Defined as any container being used beyond the original manufacturer's bottle that the chemical was shipped in.
  - Portable or working containers
  - Storage bottles that are created for distribution of smaller amounts of the chemical



# HCS Adoption of GHS – Workplace Labeling

- Employers may choose to label workplace containers
  - Use the same information from the shipping label including GHS information used to ship containers under the revised rule.
  - With label alternatives that meet the requirements for the standard (Product identifier, pictograms, symbols, or combination of information that provides hazards of the chemical)
- National Fire Protection Association (NFPA) 704 Hazard Rating and the Hazardous Material Information System (HMIS) for labeling are considered acceptable for workplace containers



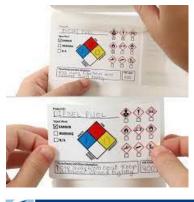
# HCS Adoption of GHS – Workplace Labeling

#### **Best Practices:**

- 1. Label your chemical secondary container with the same GHS format as the way they came into the facility
- 2. Use a dual-labeled approach to where you show both your NFPA/HMIS label alongside the GHS label









# Labeling

#### Manufacturer label

- Chemical name
- > Appropriate warnings.
- Name and address of manufacturer.

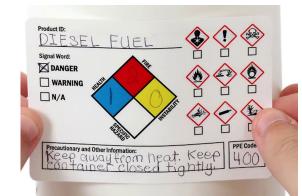




# Labeling (Continued)

#### In-house label

- Chemical name
- > Appropriate warnings
  - Remember to label ALL secondary containers immediately after putting anything into them









### NFPA Labels

- National Fire Protection Association (NFPA) Diamond
  - A system for identifying the hazards associated with chemicals.
  - Provide basic information for emergency personnel responding to a fire or spill and emergency response planners.





# **NFPA Warning Labels**

#### Example: Propane

- ≻Flammability (4)
  - Material will burn rapidly
- ≻Reactivity (0)
  - Stable
- ≻Health (1)
  - Slightly hazardous



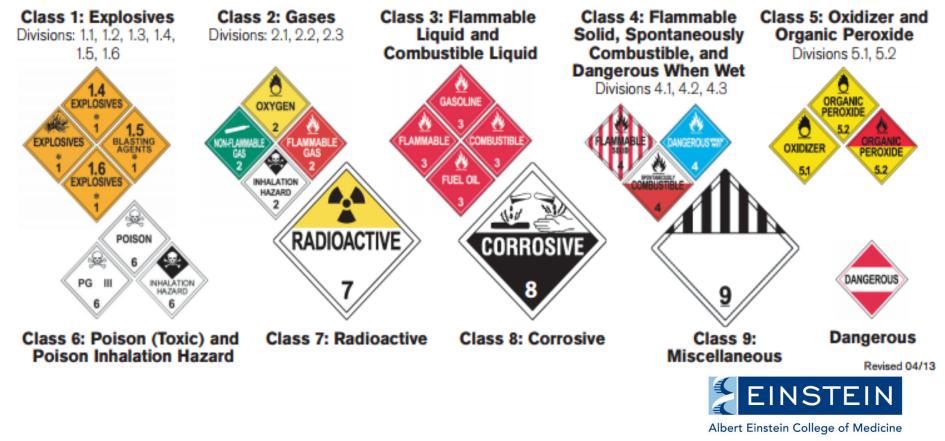
#### Example: Sulfuric Acid

- ➤ Health (3)
  - Can Cause serious or permanent injury
- ➢ Flammability (0)
- ➢ Reactivity (2)
  - Readily undergoes violent chemical changes at elevated temperatures and pressures
- ➤ Special (₩)
  - Reacts violently or explosively with water



# Warning Labels

#### Additional warning labels for identifying the hazards associated with chemicals.



### Classification Differences between HMIS/NFPA and GHS





### HCS Adoption of GHS – Safety Data Sheets (SDS)

Previously, know as Material Safety Data Sheets (MSDS).

GHS standardized the order information on SDS is presented for ease of use of employees along with improved accuracy.

Information must be provided in English. Employer may maintain copies in other languages.

Number of sections and headings increased from a 9-section format to a 16 sections. Safety Data Sheet



Section 1: Identification			
To states recomment	Treans of identification: August O assoline; 100 CL and Aviation; ASTM 100'130 Aviation Gasoline; Avgas: Avgas 100; Aviation Fuel Number: 001769 POL Annex I Category: Gasoline and Spirits		
Other means of identification:			
SDS Number:			
MARPOL Annex I Category:			
Intended Use:	Fuel		
Uses Advised Against:	All others		
Manufacturer:	SDS Information:	Emergency Health and Safety Number:	
Phillips 66 Company P.O. Box 4428	Phone: 800-762-0942	Chemtrec: 800-424-9300 (24 Hours)	
Houston, Texas 77210	Email: SDS@P66.com URL: www.Philips66.com		
10031011, 10033 17210	UNL: WWW.FIMp8007.000	Customer Service:	
		800-234-6603 Technical Information:	
		918-977-4224	
Section 2: Hazards Ident	tification		
Classified Hazards		Hazards	
H224 - Flammable liquids - Category 1		trostatic charges may be generated during handling.	
H315 – Skin corrosion/irritation Categ		같아요. 아이들에게 집에서 잘 들어진 것이 가 봐. 다귀 같은 것이다.	
H304 – Aspiration Hazard – Category 1			
H336 Specific target organ toxicity (sir H350 Carcinopenicity Category 1B	ngle exposure) - Category 3		
H350 – Carcinogenicity – Category 1B H411 – Hazardous to the aquatic enviro	orment, chimple toxicity Category 2		
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001769 - Aviation Gasoline, 100 LL Date of Issue: 04-Mar-2013 Page 1/11 Status: FINAL

# **Safety Data Sheets Elements**

- 1. Identification of the substance or mixture
- 2. Hazard Identification
- 3. Composition/information on ingredients
- 4. First aid measures
- 5. Firefighting measures
- 6. Accidental release measures
- 7. Handling and storage
- 8. Exposure controls/Personal Protection

- 9. Physical and chemical properties
- 10. Stability and reactivity
- 11. Toxicological information
- 12. Ecological information\*
- 13. Disposal information\*
- 14. Transport information\*
- 15. Regulatory information\*
- 16. Information on the preparation and revision of the SDS





# **Responsibilities of Einstein**

#### Chemical Inventory

- > Review chemical inventory
- Replace MSDS with SDS
- Dispose of expired or unwanted chemicals
- Annually review chemical inventory
  - Right-to-Know





### **Responsibilities of Einstein**

- Chemical Hygiene Plan
- Provide training to all employees with potential exposure to hazardous materials.
- Collect SDS and make them available to all employees.



#### **Chemical Hygiene Plan**

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28 June 2018

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# **Einstein - SDS Stations**

Forchheimer: Ground & 4<sup>th</sup> Floor Kennedy: 3<sup>rd</sup> Floor Price: Basement 1<sup>st</sup>, 4<sup>th</sup> & 5<sup>th</sup> Floors Van Etten Basement





# **Employee Rights**

- Request and obtain information on hazardous chemicals in your workplace.
- Be informed of hazardous chemicals you come in contact with on your job.
- Access to written material.
  - > SDS
  - > Chemical Hygiene Plan
  - > Hazard Communication Program



# NOTICE

HAZARD COMMUNICATION POLICY LOCATED HERE

 $\underline{https://intranet.einsteinmed.edu/download/?token=F3U4mtNk56VTr4YwmOleltOEedZjiAS3\%2fX4zi1xF6i4}$ 



#### **Employee Responsibilities**

- Attend training classes.
- Learn the chemical hazard by reading the SDS.
- Know the signs and symptoms of exposure.
- Know the proper personal protective equipment (PPE) to use.
- Know emergency procedures.



- Chemical
  - Approximately 600 different chemicals on campus
  - > Flammable
    - Examples: Xylene, Ethanol
  - Corrosive
    - Examples: Nitric Acid, Sodium Hydroxide
  - > Reactive
    - Examples: Sodium Metal, Sodium Azide, Picric Acid (dry)
- Biological
  - Examples: Allergens, Infectious Diseases, Toxins

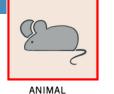






#### Physical

- > Temperature Extremes
- ➢ Electrical
- > Fire
- > Explosive
- Cryogenic Liquids
- Compressed Gasses



HAZARD



SHARP INSTRUMENT HAZARD





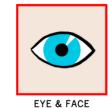
GLASSWARE HAZARD



CHEMICAL HAZARD



ELECTRICAL HAZARD



HAZARD



FIRE HAZARD



BIOHAZARD



LASER RADIATION HAZARD



RADIOACTIVE

HAZARD



EXPLOSIVE HAZARD



#### Picric Acid

 In a dry powdered state, it can become explosive when jolted (shock sensitive)





Cryogenic Liquids
 > Liquid Nitrogen



- Compressed Gas Cylinders
  - > Oxygen
  - Carbon Dioxide



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of Medicine

- Radiation
  - > Non-ionizing
    - Near Ultraviolet
      - \*Damage to skin and eyes.
    - Lasers
    - Microwave

#### > Ionizing

• Alpha, Beta, Gamma and X-rays

Generates free radicals and ion pairs in living tissue, producing damaging intermediates.

Direct hits to DNA



\*\*\* <sup>14</sup>C8

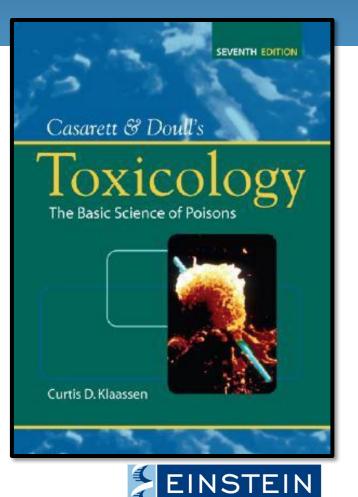


Toxicology is the study of the adverse effects of chemicals on living organisms.

Know the signs and symptoms associated with exposure to the chemicals in your workplace.

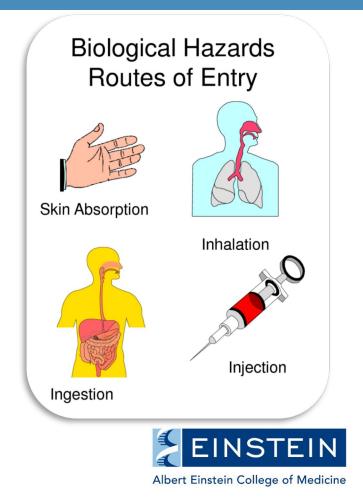
Be observant for these signs and symptoms.

Know what to do if you are exposed.





- Routes of Entry
   Inhalation (Breathing)
  - > Absorption (Direct Contact)
  - Ingestion (Eating)
  - Injection



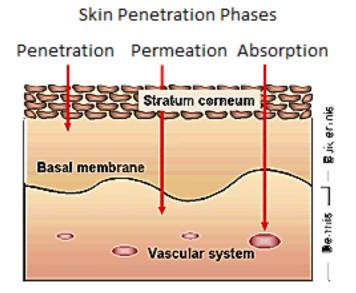
#### Inhalation

- The most common route of exposure.
- Can be in the form of a gas, vapor or dust.
- Can be deposited in the airways or absorbed through the lungs and into the blood stream.
  - Blood can then circulate the toxin to the rest of the body.



#### Absorption

- Skin acts as a barrier between the environment and the organs of the human body.
- > Skin can be attacked directly.
  - Examples: Acids and Bases
- > Skin can be penetrated.
  - Examples: Phenol, Nitrobenzene





#### Ingestion

- Rarely takes place by deliberate swallowing of toxic substances.
- Food and drinks can become contaminated by dust, mist and fumes.
- No eating or drinking in laboratories or storing food in laboratory refrigerators or cold rooms.





#### Injection

- Occurs through accidental needle sticks, puncture wounds or through broken skin/open wounds.
- May produce rapid response when injected because the chemical/toxin is introduced directly into the blood stream.





#### **Health Hazards Effects**

Acute Health Effects (Short-term):

\*Develops immediately or within minutes, hours or even days after an exposure include dizziness, skin irritation, and throat irritation.

#### Examples:

- Burns
- Headache
- Vomiting
- Nausea

Chronic Health Effects (Long-term):

\*Adverse health effect resulting from long-term exposure to a substance. Symptoms do not stop when the exposure stops.

Examples:

Asthma

Cancer

Asbestosis, Mesothelioma





- Lab Coat and Aprons
- Gloves
- Safety Glasses/Face Shields
- Hearing Protection
- Respirators





#### Lab Coats and Aprons





#### Glove Selection

- Gloves are made of many different types of materials, yet no one material type affords protection against all chemicals.
  - Latex gloves provide little to no chemical protection.





Butyl rubber Neoprene

Latex

Nitrile



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#### Ansell Glove Guide

http://www.ansellpro.com/specware/guide.asp

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	LAMINATED			NITRILE			UNSUPPORTED			SUPPORTED			POLYVINYL			NATURAL			N	EOPREN	E/	
This Information	FILM			SOL-VEX			29-865			POLYVINYL ALCOHOL PVA			CHLORIDE (Vinyl) SNORKEL			RUBBER CANNERS AND HANDLERS*			NATURAL RUBBER BLEND			
Applies Only to																						
Ansell Protective Products	BARRIER																	CHEMI-PRO*				
Glove Brands	Degradation Rating	Permeation: Breakthrough	:uo	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	Degradation Rating	Permeation: Breakthrough	Permeation: Rate	tion	Permeation: Breakthrough	iu	Degradation Rating	Permeation: Breakthrough	:uo	Degradation Rating	Permeation: Breakthrough	:uo	Degradation Rating	Permeation: Breakthrough	:uo	
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93. Nitric Acid, 70%	٤	>480	-	- 118	-	-	E	>480	-	NB.	-	-	F	345	-	NP.	-	-	G	90	-	
94. Nitric Acid, Red Furning			-	88			MR		-	NH.		-	P.			P		-	88		-	
95. Nitrobenzene		>480	E	18	-	-	NR.	_	-	G	>360	E	NR.	-	_	F	15	G	F	40	G	
96. Nitromethane, 95.5%		>480	E	F	30	F	E	60	G	G	>360	E	P	-	-	E	10	G	Ē	30	VG	
97. Nitropropane, 95.5%		>480	E	18	-	-	E	<10	F	E	>360	Ε	48	-	-	E	5	G	E	10	G	
98. Octyl Alcohol	-	-	-	E	>360	E	E	<10	Ε	G	>360	E	F	>360	Ε	E	30	VG	E	50	G	
99. Oleic Acid				E	>360	E	F	<10	G	G	60	E	F	90	VG	F	>360		G	120		
100. Oxalic Acid, saturated	-	-	-	E	>360	-	E	>480	-	P	-	-	E	>360	-	E	>360	-	E	>360	-	
101. Pad Etch 1(Ashland Chem.)	-	-	-	F	>360	-	E	>480	-	F	34	-	E	>360	-	E	>360	-	E	>360	-	
102. Palmitic Acid, saturated	-	-	-	G	30	-	E	>480	-	P	-	-	G	75	-	G	5	-	E	180	-	
103. Pentane		-	-	E	>360	Ε	G	20	F	G	>360	Ε	-88	-	-	P	-	-	ε	6	F	
104. Pentachlorophenol, 5%	-	-	-	E	>360	E	E	151	F	E	5	F	F	180	Ε	NP.	-	-	-	-	-	
105. Perchloric Acid, 60%	-	-	-	E	>360	-	E	>480	-	NR.	-	-	E	>360	—	F	>360	-	E	>360	-	
106. Perchloroethylene		>480	E	G	300	VG	NR.	-	-	E	>360	E	NR.	-	_	NR.	-	-	113	-	-	
107. Phenol		>480	E	58	-	-	E	140	F	F	>360	E	G	75	VG	E	90	-	E	180	-	
108. Phosphoric Acid, conc.		>480		E	>360		G	>480	-	NR.		-	G	>360	-	F	>360	-	G	>360		
109. PMA Glycol Ether Acetate		>480	E	E	200	F	G	37	F	E	>360	E	P.	-	-	G	13	F	G	18	F	
110. Potassium Hydroxide, 50%	-	-	-	E	>360	-	E	>480	-	A.P.	-	-	E	>360	-	E	>360	-	E	>360	-	
111. Propyl Acetate	-	-	-	F	20	G	P	-	-	G	120	VG	88	-	-	P	-	-	- P	-	-	
112. Propyl Alcohol		>480	-	E	>360	E	E	200	VG	P		-	F	90	VG	E	20	VG	E	30	VG	
113. Propylene Oxide		>480	-	NH.	-	-		-	-	G	35	G		-	-	P	-	-	Р	-	-	
114. Pyridine		>480	E	-88	-	-		-	-	G	10	F		-	-	F	10	F	P -	-	-	
115. Rubber Solvent	-	-	-	E	>360	E	E	40	F	E	>360	Ε		-	-	- NP	-	-		-	-	
116. Silicon Etch		-	-	NH.	-		E	>480	-	- NR	-	-	E.	150	-	AR.	-	-	P	-	-	
117. Skydrol hydraulic fluid	8	>480	E	-88	-	-	-88	-	-	F	-	-	-88	-	-	- NR	-	-		-	-	
118. Sodium Hydroxide, 50%	8	>480	-	E	>360	-	E	>480	-	-NB-	-	-	G	>360	-	E	>360	-	E	>360	-	
119. Stoddard Solvent		>480	E	E	>360	E	E	120	F	E	>360	E	F	360	Ε	AR.	-	-	6	10	F	
120. Styrene		>480	E	-88	-	-	- NR	-	-	G	>360	E	- 88	-	-	- NP	-	-		-	-	
121. Sulfuric Acid, 95%	ε	>480	-	-88	-	-	F	105	-	NR.	-	-	G	220	-	1.8	-	-	-88	-	-	
122. Sulfuric 47% battery acid	-	-	-	E	>360	-	E	>480	-	NR.	-	-	G	>360	-	E	>360	-	E	>360	-	
123. Tannic Acid, 65%	-	-	-	E	>360	E	E	>480	-	P -	-	-	E	>360	Ε	E	>360	-	E	>360	-	
124. Tetrachloroethene		>480	-	G	300	VG	NR.	-	-	E	>360	Ε	-88	-	-	AR.	-	-	-83	-	-	
125. Tetrahydrofuran, THF	•	>480	E	- 88	-	-	NR.	-	-	P .	90	G	-43	-	-	NP.	-	-	-43	-	-	
126. Toluene, toluol		>480	E	F	10	F	MR	-	-	G	>360	E	89	-	-	AR.	-	-	89	-	-	
127. Toluene Di-Isocyanate		>480	E	NB.	-	-	NR	-	-	G	>360	Ε	P	-	-	6	7	G	-	-	-	
128. Trichloroethylene, TCE	•	>480	E	-88	-	-	NR.	-	-	E	>360	E	-88	-	-	-NP	-	-	-83	-	-	
129. Tricresyl Phosphate, TCP	-	-	-	E	>360	E	G	<10	Ρ	G	>360	E	F	>360	8	E	45	Ε	E	>360	E	
130. Triethanolamine, 85%		-	-	E	>360	E	E	<10	G	G	>360	E	E	>360	ε	G	>360	Ε	E		-	
131. Turpentine	•	>480	E	E	30	E	NR	-	-	G	>360	Ε	P.	-	-	-NP	-	-	-83	-	-	
132. Vertrel MCA		>480	E	E	110	G	E	20	F	F	>480	E	G	13	F	6	<10	F	6	<10	F	
133. Vertrel SMT	E	<10	G	P	-	-	F	<10	Ρ	G	17	G	G	<10	F	F	<10	F	- P -	-	-	
134. Vertrel XE	ε	105	E	E	>480	E	E	47	G	F	40	VG	G	303	Ε	E	17	VG	E	43	VG	
135. Vertrel XF	3	>480	E	E	>480	E	E	>480	Ε	F	387	VG	E	>480	ε	E	337	VG	E	204	VG	
136. Vertrel XM	8	120	E	E	>480	E	E	105	Ε	F	10	G	P.	-	-	E	23	VG	E	30	VG	
137. Xylene, Xylol		>480	E	6	75	F	-NR	-	-	E	>360	E	-88	-	-	AH.	-	-	88	-	-	

#### Safety Goggles/Face Shields









Respirators
N95 particulate respirator
Filters 95% of particles 0.3 microns or greater
Not to be used for gases, vapors, or oxygen deficient areas





Hearing protection
 Earplugs

 Single use
 Earmuffs







### **Engineering Controls**

#### Ventilation

#### ≻Room air exchanges

 ~10 exchanges per hour for laboratories.







#### **Engineering Controls**

- Chemical Fume Hood Primary engineering control for containing and removing chemical gases, vapors, mist and fumes.
  - Fume Hoods must be able to remove hazardous vapors and odors from the breathing zone
  - Inspected annually
  - ➢ Face velocity of 80 120 feet per minute.
  - Sash height 12-18 inches
  - Close the sash when finished



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## **Engineering Controls**



- Do not overload the hood with extraneous equipment or chemicals.
- Not to be used as storage
- Do not evaporate chemicals
- A cluttered hood can compromise the airflow patterns and negate the hoods safety features



#### Order only what is needed

- > Each lab is required to store their own materials
- Make sure chemicals are labeled properly
  - As well as all secondary containers
- Avoid floor and top shelf storage
  - Nothing above eye level
- Store flammables in appropriate cabinets.
  - 15 gallons (56 liter) limit per lab.
  - 25 gallons (94 liter) limit per lab with sprinkler system (Golding, MRRC & Price Building)



- Separate acids and bases
  - > Organics vs. Inorganics

• Examples:

- Hydrochloric Acid (HCl) and Sodium Hydroxide (NaOH)
- Sulfuric Acid (H<sub>2</sub>SO4) and Formic Acid (HCOOH)
- Perchloric acid must be stored in glass containers separated from organic materials.
- Segregate oxidizers from organic liquids.
  - Examples: Hydrogen peroxide, Permanganates, Nitrate compounds



OH-

 Nitric acid must be isolated from other acids and bases.





- Dispose of used / expired chemicals via EH&S.
  - Examples: Ethyl Ether, 2-Propanol
- May form explosive peroxides after one year. Exposure to light and/or air significantly increases the rate of peroxide formation.
  - Complete the required peroxide tests and forms







- Acid Storage
  - Store in plastic secondary containers, not directly on metal shelves



#### Flammable Storage

> Do not store cardboard or

Styrofoam





- Chemical disposal is free.
- Use the proper waste container.
  - Examples: Bio Bins, Sharps, Ethidium Bromide gels, Radioactive, Office waste, Recycle waste.
- Rinse empty bottles and deface label.
- Label all chemical waste.



# Hazardous Materials Spill Cleanup



#### MINOR CHEMICAL SPILLS CLEANUP PROCEDURES:

- Do not clean up the spill yourself, unless you know what spilled and how to properly clean it up.
- Immediately, notify others in the area that a spill has occurred.
- Turn off ignition and heat sources.
- Prevent others from coming in contact with the spilled chemical(s).
- Wear proper PPE (i.e., goggles, gloves)
- Use the appropriate material to confine or contain the spill to avoid spreading
- Absorb inorganic acids and bases and neutralize.
- Sweep up the absorbed spill from the outside toward the middle. Scoop up and deposit in a leak-proof container. Label and dispose of the container through the hazardous waste management program
- Collect the residue, place in a container, and dispose as hazardous waste.

NEVER PLACE HAZARDOUS MATERIALS OR SPILL CLEANUP MATERIALS IN THE BIOHAZARD BIN OR REGULAR TRASH.

- Clean spill area with water.
- Report spill to EH&S at X4150





#### Hazardous Materials - Spill Cleanup

#### MAJOR CHEMICAL SPILLS CLEANUP PROCEDURES:

- If the chemical spill presents an immediate danger; turn off ignition and heat sources, evacuate all personnel, exit the room/area, and close the door behind you.
- Attend to anyone who may have been contaminated or injured and remove them from exposure.
- Instruct contaminated person to remove any contaminated clothing and wash the affected area for at least 15 minutes.
- Use the safety shower if necessary know the location of the nearest safety shower and eye wash station in your work area.
- Notify EH&S at X4150 and Security at X4111

NEVER PLACE HAZARDOUS MATERIALS OR SPILL CLEANUP MATERIALS IN THE BIOHAZARD BIN OR REGULAR TRASH.







#### **Emergency Procedures**

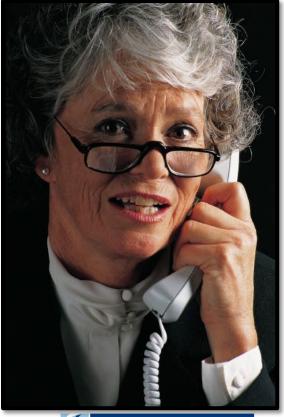
Remain Calm.

Protect yourself and your colleagues from further harm.

Notify EH&S, Security, and/or Fire Department as appropriate.

x4150 x4111 911

Wash any chemical contact for 15 minutes. Seek medical attention if needed.





### Summary

- SDS are available to anyone.
- Understand the new GHS and label requirements
- Understand the hazards of the chemicals you work with.
- Store chemicals properly.
- Wear the appropriate PPE.
- Always use a chemical fume hood.
- Keep the work area clean.
- Wash affected area for at least 15 minutes.
- Call EH&S with any questions.



### Hazardous Communication Sign in Sheet and Quiz



https://einsteinmed.co1.qualtrics.com/jfe/form/SV\_0rluXDP135pIhW6

