Selected Topics of New Frontiers in Computational Systems Biology

Yinghao Wu Albert Einstein College of Medicine

Department of systems and computational biology

Outline

Modeling and Simulating Larger Molecular Systems with Longer Time Scales

* Coarse-grained simulation

* <u>Multi-scale</u> modeling

Extend the Applications of Structural Biology to Newly Emerging Fields

* Fold proteins by information from evolution

Simulating crowded <u>cellular environment</u>

* Combining structural biology with systems biology

Summary and Perspective

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Background: Why Coarse-graining (CG) ?

All atom:All atom:time step of 1-2 fs,each particle represent one atomtime frame sampled ~100 nsMaximally can handle n-atom system

Reduction in degrees of freedom
Fast frequency movements removed
Smoother potential surface
Longer time steps can be taken
Microsecond simulations possible

Coarse-grained: time step of 20-50 fs, time frame sampled ~1 µs Coarse-grained: each particle represent m atom Maximally can handle mn-atom system

Background: How Coarse-graining (CG) ?

How to give a simplified representation for biomolecule with reasonable accuracy (CG model).

How to get the correct interactions between different simplified sites within the models (Force Mapping).

MATINI is a software to do CG simulations.

Coarse-grained Model of Phosphor-lipid



Coarse-grained Model of Protein Side-chain: Mapping Sites



Coarse-grained Model of Protein Main-chain: Degrees of Freedom



Comparison between All Atom Model and Coarse-grained Model



Bonded Interactions



$$V_{\rm b} = \frac{1}{2} K_{\rm b} (d_{ij} - d_{\rm b})^2$$

$$V_{a} = \frac{1}{2} K_{a} [\cos(\varphi_{ijk}) - \cos(\varphi_{a})]^{2}$$
$$V_{d} = K_{d} [1 + \cos(n\psi_{ijkl} - \psi_{d})]$$
$$V_{id} = K_{id} (\psi_{ijkl} - \psi_{id})^{2}$$

- Bonded interactions are described by the following set of potential energy functions acting between bonded sites i, j, k, and I with equilibrium distance, angle, and dihedral.
- The force constants K are generally weak, inducing flexibility of the molecule at the coarse-grained level mimicking the collective motions at the fine-grained level. The bonded potential is used for chemically bonded sites and the angle potential to represent chain stiffness.
- The improper dihedral angle potential is used to prevent out-of-plane distortions of planar groups. Proper dihedrals are used to impose secondary structure of the peptide backbone.c

Calculating bonded force constants: 1) deriving distributions from PDB

- The bond lengths, bond angles, dihedral angles, and their respective force constants, collectively referred to as the bonded parameters, were obtained from distributions derived from the PDB.
- Chose a representative subset of approximately 2000 proteins from the PDB as the basis set for our parametrization.
- The secondary structure of every residue of these proteins was determined using the program DSSP.
- Using the center of mass of the atoms representing each coarse-grained bead, calculated the distributions of the bond lengths, bond angles, and dihedral angles.



Calculating bonded force constant: 2) deriving force constants against distributions of PDB

- After the distributions were obtained, simulations were performed on short test peptides, with different sequences and secondary structure characteristics.
- All of the bonded parameters were optimized by matching the PDB distributions of the bonds angles and dihedrals with the distributions obtained from the simulations, using an iterative procedure.

Table 3. Backbone Bonded Parameters

backbone	d _{BB} (nm)	K _{BB} (kJ nm ^{−2} mol ^{−1})	θ _{BBB} (deg)	<i>K</i> _{BBB} (kJ mol⁻¹)	ψ_{BBBB} (deg)	<i>K</i> _{BBBB} (kJ mol ^{−1})
helix	0.35	1250	96 ^a	700	60	400
coil	0.35	200	127	25		
extended	0.35	1250	134	25	180	10
turn	0.35	500	100	25		
bend	0.35	400	130	25		

^{*a*} $\theta_{BBB} = 98^{\circ}$ when Proline is in the helix; $K_{BB} = 100 \text{ kJ mol}^{-1}$.

Table 5. Equilibrium Angles, Improper Dihedral Angles and Force Constants for Side Chains

side chain	θ (deg)	<i>K</i> (kJ mol ^{−1})				
θ_{BBS} (all)	100	25				
θ_{BSS} (Lys, Arg)	180	25				
θ_{BSS} (His, Tyr, Phe)	150	50				
θ _{BSS} (Trp)	90, 210	50, 50				
side chain	ψ (deg)	K (kJ rad ⁻² mol ⁻¹)				
ψ_{BSSS} (His, Tyr, Phe)	0	50				
$\psi_{\rm BSSS}$ (Trp)	0, 0	50, 200				

Non-bonded Interactions: Lennard-Jones Potential

$$V_{\text{Lennard-Jones}}(r_{ij}) = 4\varepsilon_{ij} \left[\left(\frac{\sigma_{ij}}{r_{ij}} \right)^{12} - \left(\frac{\sigma_{ij}}{r_{ij}} \right)^6 \right]$$

- The strength of the interaction, determined by the value of the well depth εij depends on the interacting particle types.
- The value of ε ranges from εij) 5.6 kJ/mol for interactions between strongly polar groups to εij) 2.0 kJ/mol for interactions between polar and apolar groups mimicking the hydrophobic effect.
- The effective size of the particles is governed by the LJ parameter: σ) 0.47 nm for all normal particle types.

Non-bonded Interactions: Interaction Matrix

TABLE 1: Interaction Matrix^a

		Q				Р				N				С					
	sub	da	d	a	0	5	4	3	2	1	da	d	a	0	5	4	3	2	1
Q	da	0	0	0	II	0	0	0	Ι	Ι	Ι	Ι	Ι	IV	V	VI	VII	IX	IX
Ĩ	d	0	Ι	0	II	0	0	0	Ι	Ι	Ι	III	Ι	IV	V	VI	VII	IX	IX
	a	0	0	Ι	II	0	0	0	Ι	Ι	Ι	Ι	III	IV	V	VI	VII	IX	IX
	0	II	II	II	IV	Ι	0	Ι	II	III	III	III	III	IV	V	VI	VII	IX	IX
Р	5	0	0	0	Ι	0	0	0	0	0	Ι	Ι	Ι	IV	V	VI	VI	VII	VIII
	4	0	0	0	0	0	Ι	Ι	II	II	III	III	III	IV	V	VI	VI	VII	VIII
	3	0	0	0	Ι	0	Ι	Ι	II	II	II	II	II	IV	IV	V	V	VI	VII
	2	Ι	Ι	Ι	II	0	II	II	II	II	II	II	II	III	IV	IV	V	VI	VII
	1	Ι	Ι	Ι	III	0	II	II	II	II	II	II	II	III	IV	IV	IV	V	VI
Ν	da	Ι	Ι	Ι	III	Ι	III	II	II	II	II	II	II	IV	IV	V	VI	\mathbf{VI}	VI
	d	I	III	Ι	III	Ι	III	II	II	II	II	III	II	IV	IV	V	VI	VI	VI
	a	Ι	Ι	III	III	Ι	III	II	II	II	II	II	III	IV	IV	V	VI	VI	VI
	0	IV	IV	IV	IV	IV	IV	IV	III	III	IV	IV	IV	IV	IV	IV	IV	V	VI
С	5	V	V	V	V	V	V	IV	IV	IV	IV	\mathbf{V}	V						
	4	VI	VI	VI	VI	VI	VI	V	IV	IV	V	V	V	IV	IV	IV	IV	V	V
	3	VII	VII	VII	VII	VI	VI	V	V	IV	VI	VI	VI	IV	IV	IV	IV	IV	IV
	2	IX	IX	IX	IX	VII	VII	VI	VI	V	VI	VI	VI	V	V	V	IV	IV	IV
	1	IX	IX	IX	IX	VIII	VIII	VII	VII	VI	VI	VI	VI	VI	V	V	IV	IV	IV

^{*a*} Level of interaction indicates the well depth in the LJ potential: O, $\epsilon = 5.6$ kJ/mol; I, $\epsilon = 5.0$ kJ/mol; II, $\epsilon = 4.5$ kJ/mol; III, $\epsilon = 4.0$ kJ/mol; IV, $\epsilon = 3.5$ kJ/mol; V, $\epsilon = 3.5$ kJ/mol; VI, $\epsilon = 2.7$ kJ/mol; VII, $\epsilon = 2.3$ kJ/mol; VIII, $\epsilon = 2.0$ kJ/mol; IX, $\epsilon = 2.0$ kJ/mol. The LJ parameter $\sigma = 0.47$ nm for all interacion levels except level IX for which $\sigma = 0.62$ nm. Four different CG sites are considered: charged (Q), polar (P), nonpolar (N), and apolar (C). Subscripts are used to further distinguish groups with different chemical nature: 0, no hydrogen-bonding capabilities are present; d, groups acting as hydrogen bond donor; a, groups acting as hydrogen bond acceptor; da, groups with both donor and acceptor options; 1–5, indicating increasing polar affinity.

Deriving the parameters in the matrix by the same procedure that was used to derive the parameters for the bonded interactions

Non-bonded Interactions: Coulombic Potential

$$V_{\rm el} = \frac{q_i q_j}{4\pi\varepsilon_0 \varepsilon_{\rm rel} r_{ij}}$$

charged groups (type Q) bearing a charge q interact via a Coulombic energy function with a relative dielectric constant εrel) 15 for explicit screening.

SNARE-mediated membrane fusion



- a) The two membranes are in the vicinity of each other but the SNAREs are not yet in contact.
- b) SNARE complexes start zipping from the amino-terminal end, which draws the two membranes further towards each other.
- c) Zipping proceeds, causing increased curvature and lateral tension of the membranes, exposing the bilayer interior.
- d) The highly unfavorable void space at the membrane junction in (c) causes the establishment of contacts between the distal membrane leaflets.
- e) The lateral tension in the trans-bilayer contact area induces membrane breakdown, yielding a fusion pore.
- f) The fusion pore expands and the membrane relaxes.

SNARE-mediated membrane fusion: initial set-up



SNARE-mediated membrane fusion: stalk formation during simulation



SNARE-mediated membrane fusion: four stages of fusion



A) I: Stalk, II : inverted micelle intermediate (IMI; single SNARE complex), III: Hemifusion-diaphragm (single SNARE complex), IV: fusion. B) Cross-sections of the fusion plane showing the transition from stage I (stalk) to stage II (inverted micelle) in more detail. The stalk (664 ns), shows a worm-like expansion (740 ns) and eventually, after bending (920 ns), forms a closed ring that encapsulates exterior solvent forming an inverted micelle (968 ns).

SNARE-mediated membrane fusion: conclusions

- SNARE complex is driven towards the perimeter of the fusion region
- SNARE complexes induce stalk formation and actively opens fusion pores
- Intrinsic properties of trans-membrane regions facilitate stalk formation
- SNARE complexes release their energy throughout the whole fusion process up to the expansion of the fusion pore

Beyond MARTINI



Domain-based Coarse-graining

Beyond MARTINI



Coarse-graining the molecule into functional units

Beyond MARTINI

Optimizing the number of coarse-grained sites in different components of large bio-molecular complexes



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A multi-scale nature of biological systems







The multi-scale model of cell population growth



Moving Sets



Lattice-based Simulations



Simulation results of normal cells



Simulation results of tumor cells



Conclusions

The multi-scale simulations show the dynamics of cell populations and their responses to nutrient shortage.

Tumour cells take over the space originally occupied by normal cells and keep growing in the hypoxic areas, owing to their higher rates of oxygen consumption.





Cadherin-mediated Cell Adhesion


Molecular Structure of Type-I Classical Cadherin



AUC can only measure 3D binding affinity

Cis binding too weak to be measured

Jin X, Harrison O, et. al., Structure (2011)

From 3D to 2D Binding Affinity: Theory and Simulations



Cadherin clustering: Anisotropic Lattice MC Simulation



Cadherin clustering: Simulation versus Experiements



Beyond the molecular level



Molecular Level

Subcellular Level

Cellular Level

Individual-based Cellular Simulation

Voronoi tessellation as the representation of cell geometry



> Hamiltonian function describing cell-cell interactions

$$\boldsymbol{U}_{Total} = -\sum_{ij} \gamma(\tau_i, \tau_j) \boldsymbol{P}_{ij} + \lambda_A \sum_i (A_i - A_i^0)^2 + \lambda_P \sum_i (P_i - P_i^0)^2$$

Junction simulation

> Use simulated annealing to search the energy minimal state

From Cadherin Binding Specificity to Cell Aggregation



Cells Expressing N-Cadherin

 $100 \ \mu m$



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Protein Folding



Proteins don't have a "folding problem"... ... it's we humans that do!



Physical-based potential

Knowledge-based potential

Physical-based Potential





2F4K 1.3 Å 2.8 µs

125 µs

BBL 2WXC 4.8 Å 29 µs



Villin

325 µs





A-repressor 643 μs 1LMB 1.8 Å 49 µs

David Shaw Group

707 µs

Knowledge-based Potential



The problem of knowledge-based potential to study protein folding



Can we get insights from other disciplines ?

Evolution Patterns

Correlated mutations carry information about distance relationships in protein structure.



These patterns can be used to predict residue contacts in proteins



Predicted contacts can be transferred into potential to fold proteins



Results



Conclusions

Three-dimensional structure from coevolution patterns—why does it work?

- Precise information in the evolutionary sequence record.
- Growth in sequence databases from massively parallel sequencing.
- Reduction of conformational search space by cooperative probability models.

Future Directions



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Background

There are high concentrations of macromolecules in living cells



The crowding effect can make molecules in cells behave in radically different ways than in testtube assays.

Consequently, measurements of the properties of enzymes or processes in metabolism that are made in the laboratory (*in vitro*) in dilute solutions may be different by many orders of magnitude from the true values seen in living cells (*in vivo*).

The study of biochemical processes under realistically crowded conditions is very important, since these conditions are a ubiquitous property of all cells and crowding may be essential for the efficient operation of metabolism.

Background

Crowding effects alter molecular properties



- The size of the crowding effect depends on both the molecular mass and shape of the molecule involved.
- This excluded volume effect alters the rates and equilibrium constants of molecular reactions.
- This effect alters dissociation constants of macromolecules, such as when multiple proteins come together to form protein complexes, or when DNA-binding proteins bind to their targets in the genome, or enzyme reactions involving small molecules in solvent.

Model Setup

					1				1	1			1			
	Name	Mw	#		Name	Mw	#		Name	Mw	#		Name	Mw	#	
٠	Adk	24	14		GapA	142	10		PanB	140	2	۲	Suc	142	4	
0	AhpC	187	7		GlnA	621	1		Pgk	41	26	-	Tig	48	9	
	Asd	80	4	-	GltD	94	3	3	Pnp	190	3		TpiA	54	5	
٠	Вср	11	8		GlvA	91	15	6	Ppa	116	9	-	Tsf	61	12	Bern March 199
٠	CspC	7	72		GpmA	55	4	٠	PpiB	18	7		TufA	84	181	
-	CysK	64	13		Hns	5	7	٠	PurA	94	4	٠	Upp	45	11	
	DapA	125	2		Hun	15	12		PurC	42	7	•	UspA	31	7	
٠	DnaK	41	11		Inda	15	12	۲	Pyr	308	3	Q at	50S	1,355	10	
٠	Efp	20	14	-	ICUA ILuC	92	43	۰	RpiA	46	3		305	788	10	
٠	Eno	91	18	-	Nut	54	18	4	Rpo	260	4		tRNA-C	24	37	ALC: NOT DEC
-	Fba	78	6		Man	65	13	۰	SerC	79	11	4	tRNA-O	24	37	
*	Frr	21	7		MetE	84	213		SodA	46	13	4	tRNA-F	25	37	
-	FusA	69	22		Мор	845	2		SodB	42	9		GFP	26	8	

Skolnick Group & Elcock Group

Stochastic Dynamics: Equation of Motion

$$\begin{split} \overline{F}_{ij} &= -\left(\partial V_{ij} / \partial d_{ij}\right) \times \left(\frac{d(d_{ij})}{dx}, \frac{d(d_{ij})}{dy}, \frac{d(d_{ij})}{dz}\right) \\ \hline Forces \\ \hline \textbf{Langevin Equation} \\ \widehat{\Gamma}_{i-E}^{f} &\approx \overline{v}_{i} + \sum_{j \in (d_{ij} < (R_{i} + R_{j}))} \widehat{\Gamma}_{ij}^{f} \times (\overline{v}_{i} - \overline{v}_{j}) = \sum_{j \in (d_{ij} < (R_{i} + R_{j}))} \overline{F}_{ij} + f_{i}(t) \\ \hline \textbf{Molecule-environment friction} \\ \hline \textbf{Inter-molecule friction} \\ \widehat{\Gamma}_{ij}^{f} &= \eta \widehat{I} \\ \widehat{\Gamma}_{ij}^{f} &= \eta \widetilde{I} \\ \widehat{\Gamma}_{ij}^{f}$$

Simulation Trajectory



Simulation Results: Diffusion



Simulation Results: Stability



- The excluded-volume (crowding) effect experienced by proteins significantly stabilizes their folded states relative to their unfolded states.
- The effect is counterbalanced by the more favorable energetic interactions engaged in by the unfolded state conformations.
- These results suggest that differences between the in vitro and in vivo thermodynamic stabilities will vary significantly with the identity of the protein.

Simulation Results: Binding



Excluded-volume crowding effect, which alone stabilizes dimers relative to separated monomers is largely cancelled by the more favorable energetic interactions that the monomers form with the cytoplasm constituents.

Summary and Conclusions

- Developing computational models of intracellular environments is the route to understanding differences between biomolecular behavior observed in vitro and in vivo.
- The simulation model successfully describes the relative thermodynamic stabilities of proteins measured in E. coli, and shows that effects additional to the "crowding" effect must be included in attempts to understand macromolecular behavior in vivo.
- the simulation approach offers a potentially important complement to experimental techniques and provides a vivid illustration of molecular behavior inside a biological cell

Future Directions

The internal conformational variations of macromolecules or transitions between different states in molecular motors.

Effects from cellular compartmentation and cytoskeleton.

Real biological applications.

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***** Summary and Perspective

Background: Protein-protein interactions (PPIs)



PPIs are involved in many biological processes:

- Signal transduction
- Protein complexes or molecular machinery
- Protein carrier
- Protein modifications (phosphorylation)

—

PPIs prediction: predict if two proteins interact with each other or not, given their sequences.

Background: experimental methods

The first comprise and 'atomic observation' in which the protein interaction is detected using, for example, X-ray crystallography. These experiments can yield specific information on the atoms or residues involved in the interaction.

- The second is a 'direct interaction observation' where protein interaction between two partners can be detected as in a two-hybrid experiment.
- At a third level of observation, multi-protein complexes can be detected using methods such as immuno-precipitation or massspecific analysis. This type of experiment does not unveil the chemical detail of the interactions or even reveal which proteins are in direct contact but gives information as to which proteins are found in a complex at a given time.
- The fourth category comprises measurements at the cellular level, where an 'activity bioassay' is used to observe an interaction; for example, proliferation assays of cells by a receptor-ligand interaction.

Background: PPIs databases

- Recent advances in proteomics technologies such as two-hybrid, phage display and mass spectrometry have enabled us to create a detailed map of biomolecular interaction networks. Initial mapping efforts have already produced a wealth of data. As the size of the interaction set increases, databases and computational methods will be required to store, visualize and analyze the information in order to effectively aid in knowledge discovery.
- For the protein-protein interactions, there are many websites can be reached, here I just show several.
 - <u>BIND</u> (Interaction Network Database)
 - <u>DIP</u> (Database of Interacting Proteins)
 - Protein-Protein Interaction Server
 - Protein-Protein Interface

Problem: too much information, many false positives and hard to prove.



Idea: what computational structural biologists can do?



IS-score = 0.49, $P = 1.5 \times 10^{-5}$

IS-score = 0.43, $P = 1.1 \times 10^{-4}$

IS-score = 0.40, $P = 3.2 \times 10^{-4}$ RMSD = 2.8 Å, $f_{res} = 86\%$, $f_{con} = 85\%$ RMSD = 2.2 Å, $f_{res} = 83\%$, $f_{con} = 61\%$ RMSD = 3.1 Å, $f_{res} = 95\%$, $f_{con} = 63\%$

Although protein complexes only share remote structural similarity, they sometimes tend to bind with each other with conservative interfaces.

Can we use this structural information to help PPI prediction?

Method: prediction procedure



Honig Lab
Method: model evaluation



Prediction results



Prediction results



Prediction results



Summary and Conclusions

- It has been demonstrated that even distantly related proteins often use regions of their surface with similar arrangements of secondary structure elements to bind to other proteins, considerably expanding the number of putative PPIs that can be identified.
- The study suggests the ability to add a structural 'face' for a large number of PPIs, and that structural biology can have an important role in molecular systems biology.
- Different methods with diverse backgrounds can be integrated to effectively remove false positives in PPI interactome.

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